



Active Colloidal Swimmers as building blocks for active matter

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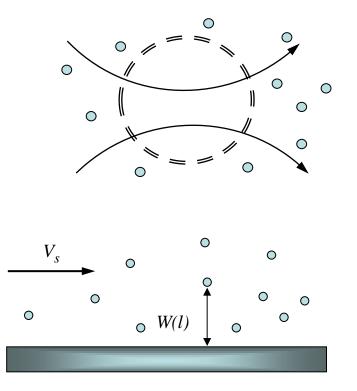
Active Colloids

Questions we would like to address:

- How can we make colloids active?
- How will their behaviour differ from passive colloids?
- What happens when we put them together and they interact?
- Could we design dead swimmers who can exhibit the same behaviour as living swimmers?

Non-equilibrium Transport: Phoretic Motion

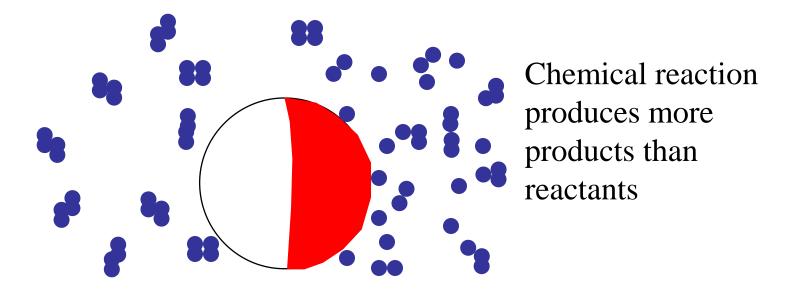
- Electrophoresis Smoluchowski (1903)
- Thermophoresis Ludwig (1856), Soret (1879)
- Osmophoresis Sackmann *et al* (1999)
- Diffusiophoresis Derjaguin *et al* (1961), Anderson & Prieve (1984)



Phoretic Transport is Force-Free

Propulsion via Self-Phoresis

e.g. self-diffusiophoresis



Propulsion (repulsive interaction)

Particle half-coated with catalyst

Analysis of the Motion

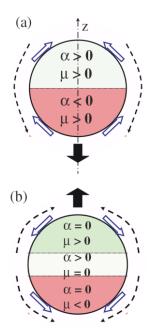
• Slip Velocity $\mu = k_{\rm B}T\lambda^2/\eta$ $\lambda_D^2 = \int_0^\infty dl \ l \left[1 - \mathrm{e}^{-W(l)/k_{\rm B}T} \right]$ $\mathbf{v}_{s}(\mathbf{r}_{s}) = \mu(\mathbf{r}_{s})(\mathbf{I} - \mathbf{nn}) \cdot \nabla c(\mathbf{r}_{s})$ • Source at the Surface $D\nabla^2 c = 0$ mobility $-D\mathbf{n} \cdot \nabla c(\mathbf{r}_s) = \alpha(\mathbf{r}_s)$ activity

Propulsion via Self-Phoresis

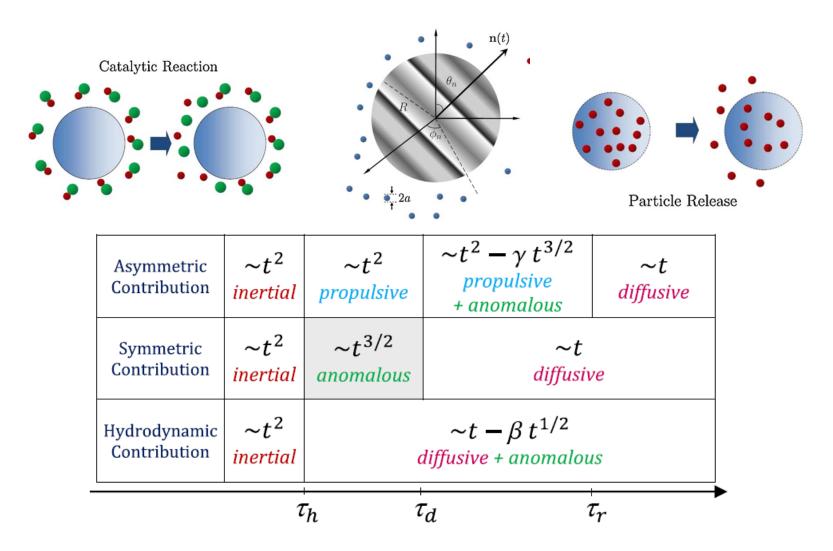
• Swimming Velocity

 $V \sim \alpha \mu / D$

- Design-related Questions
 - Symmetry breaking
 - Relative importance of activity/mobility
 - Effect of Geometry
- Generic for Any Phoresis

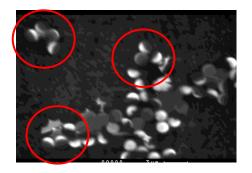


Stochastic Dynamics of Phoretic Swimmers



R. Golestanian (2009)

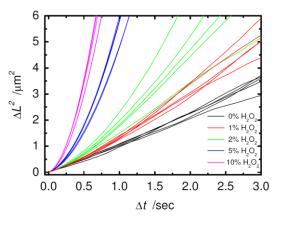
Experiments on Self-Phoresis

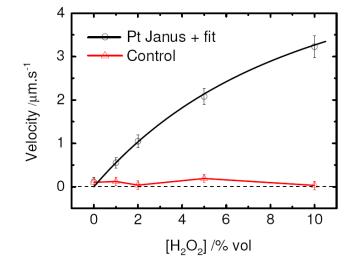


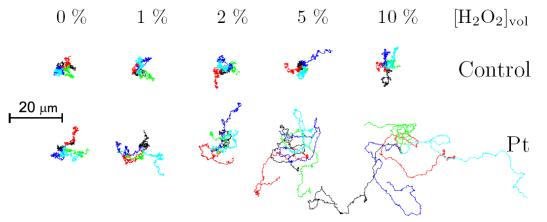
half-Pt coated spherical PS beads in aqueous hydrogen peroxide solution



Ebbens et al (2010)







Howse et al (2007)

Outline of My Talk

- I. Self-Assembled Active "Molecules" in Dilute Heterogeneous Mixtures of Symmetric Colloids
- II. Collective Chemotaxis of Asymmetric Active Colloids: the appearance of Debye screening, and various forms of instability
- III. Self-Organised Swarms in light-activated thermally active symmetric colloids that exhibit phototaxis

<u>Part I</u>

Chemical Activity

... Self-Assembled Active 'Molecules'' ...

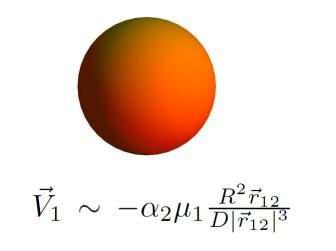
Active Colloidal Molecules

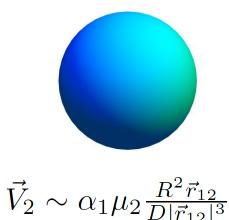
Source or Sink of Chemicals:

 $C(r) \sim \alpha R^2 / (Dr)$

Analogous to Coulomb or Gravitational potential

Two active colloids: it's complicated!





Action \neq Reaction

R. Soto & R. Golestanian (2014)

Active Self-Assembly

Brownian dynamics simulation:

$$\frac{d\vec{r}_{i}}{dt} = \sum_{k \neq i} \frac{\alpha_{k} \mu_{i} R^{2}}{6\pi D} \frac{\vec{r}_{ik}}{|\vec{r}_{ik}|^{3}} + \vec{\xi}_{i}(t) = V_{0} \sum_{k} \widetilde{\alpha}_{k} \widetilde{\mu}_{i} \frac{\sigma^{2} \vec{r}_{ik}}{|\vec{r}_{ik}|^{3}} + \vec{\xi}_{i}(t)$$

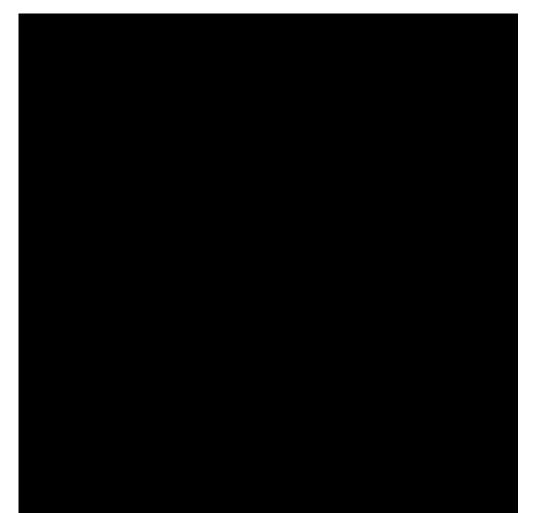
+excluded volume effect.

Colloidal molecules spontaneously form via self-assembly, and exhibit different types of activity depending on symmetry

Two types of colloids for simplicity, with different valences:

$$A \quad \bigcirc \qquad \widetilde{\alpha}_A \ge 1 \quad \widetilde{\mu}_A \ge 0$$
$$B \quad \bigcirc \qquad \widetilde{\alpha}_B = \widetilde{\mu}_B = -1$$





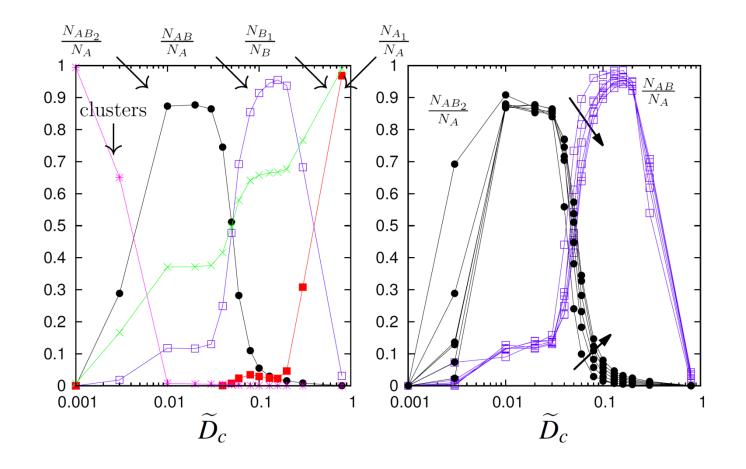
 $B \quad \bigcirc$ $\widetilde{\alpha}_B = \widetilde{\mu}_B = -1$

$$A \bullet$$
$$\widetilde{\alpha}_A = 3$$
$$\widetilde{\mu}_A = 0$$

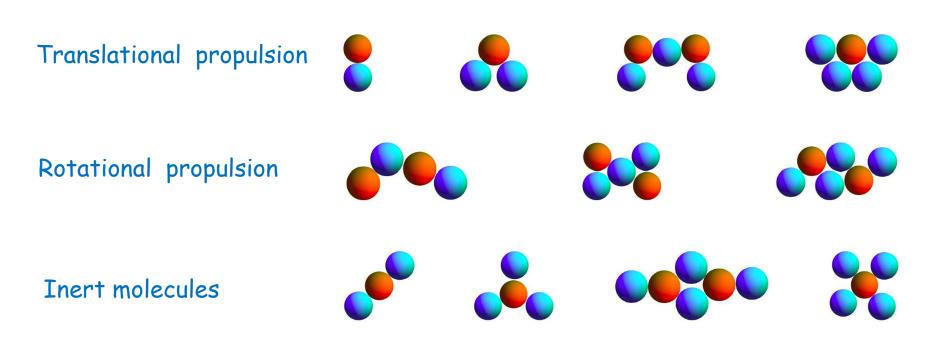
Molecular Populations vs Temperature

$$A \quad \bullet \quad \widetilde{\alpha}_A = 3 \quad \widetilde{\mu}_A = 0$$
$$B \quad \bullet \quad \widetilde{\alpha}_B = \widetilde{\mu}_B = -1$$

Stable molecules are not necessarily "charge neutral"

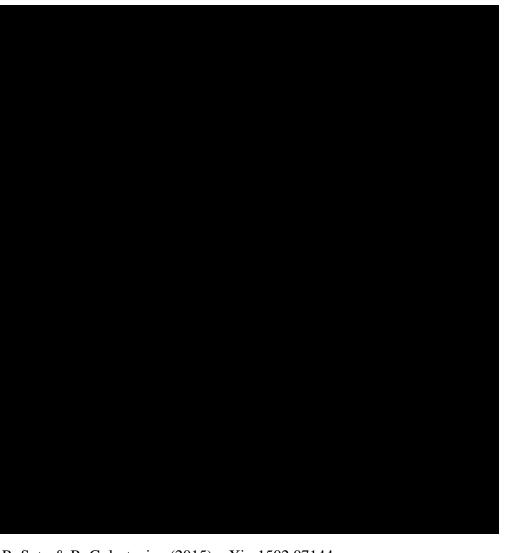


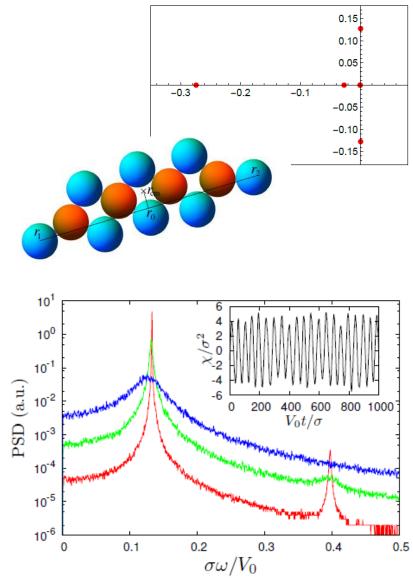
Tabulating Active Molecules



3D Structure determines Function, like proteins

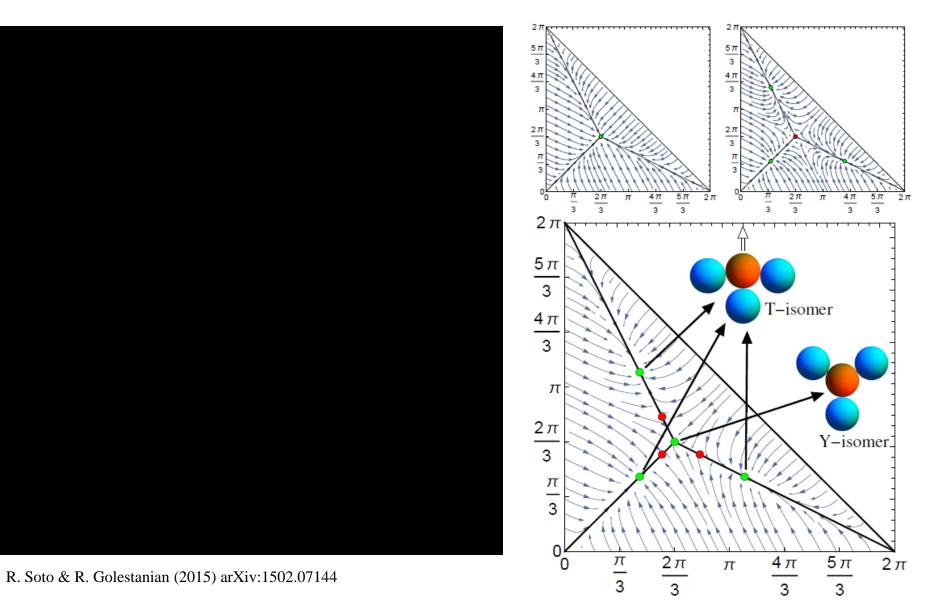
Dynamic Function: Spontaneous Oscillation





R. Soto & R. Golestanian (2015) arXiv:1502.07144

Dynamic Function: Run & Tumble



Dynamic Function: Beating & Swimming



self-assembled "artificial sperm"

... and its history ...

R. Soto & R. Golestanian (2015) arXiv:1502.07144

Part II

Chemical Activity

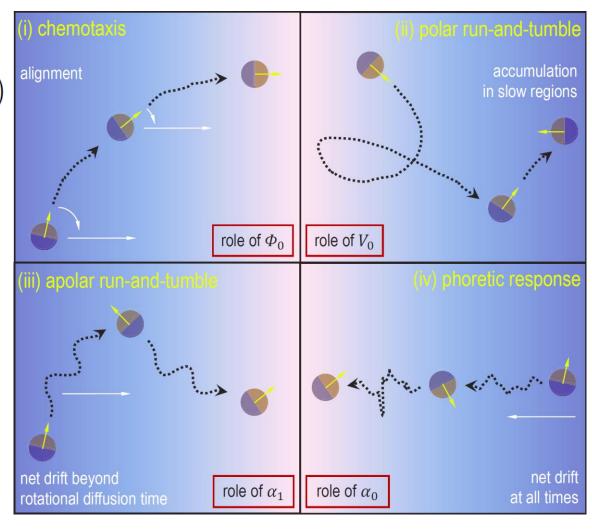
... Collective Chemotaxis of Janus Colloids...

Response to Nonuniform Chemical Profile

 $S \to P$ $\kappa(s) \equiv \kappa_2 \kappa_1 s / (\kappa_2 + \kappa_1 s)$

Four Channels with different origins

$$\boldsymbol{\omega} = \Phi_0(\sigma, \mu_p, \mu_s) \, \hat{\mathbf{n}} \times \boldsymbol{\nabla} s,$$



 $\mathbf{v} = V_0(s)\hat{\mathbf{n}} - \alpha_0 \nabla s - \alpha_1 \hat{\mathbf{n}} \hat{\mathbf{n}} \cdot \nabla s$

Parameters Give Design Rules

$$\begin{split} \Phi_0 &= -\frac{3\mu_{s1}}{4R} - \frac{\kappa_1}{60D_p} \left(5\mu_{p1}\sigma_0 + 2\mu_{p2}\sigma_1 - \mu_{p1}\sigma_2 \right) \\ V_0 &= \frac{\kappa_1 s_b}{15D_p} (5\sigma_1\mu_{p0} + 2\sigma_2\mu_{p1} - \sigma_1\mu_{p2}), \\ \alpha_0 &= -(\mu_{s0} + \frac{1}{10}\mu_{s2}) - \frac{\kappa_1 R}{10D_p} (\sigma_0\mu_{p2} - \frac{2}{9}\sigma_1\mu_{p1} - 2\sigma_2\mu_{p0} \\ &+ \frac{1}{35}\sigma_2\mu_{p2}), \\ \alpha_1 &= -\frac{1}{10}\mu_{s2} - \frac{\kappa_1 R}{30D_p} (10\sigma_0\mu_{p0} + \sigma_0\mu_{p2} + 2\sigma_1\mu_{p1} - 2\sigma_2\mu_{p0} \\ &+ \frac{29}{35}\sigma_2\mu_{p2}). \\ \Omega_0 &= -\frac{3\mu_{p1}}{4R}, \ \beta_0 = -(\mu_{p0} + \frac{1}{10}\mu_{p2}), \ \beta_1 = -\frac{1}{10}\mu_{p2} \end{split}$$

Many-Colloid Theory

$$\frac{\mathrm{d}\mathbf{r}_{\alpha}}{\mathrm{d}t} = V_{0}(s)\mathbf{\hat{n}}_{\alpha} - \alpha_{0}\nabla s - \alpha_{1}\mathbf{\hat{n}}_{\alpha}\mathbf{\hat{n}}_{\alpha}\cdot\nabla s + \beta_{0}\nabla p +\beta_{1}\mathbf{\hat{n}}_{\alpha}\mathbf{\hat{n}}_{\alpha}\cdot\nabla p + \sqrt{2D}\mathbf{f}_{\alpha}^{r}(t), \frac{\mathrm{d}\mathbf{n}_{\alpha}}{\mathrm{d}t} = \Phi_{0}(\mathbf{\hat{n}}_{\alpha}\times\nabla s)\times\mathbf{\hat{n}}_{\alpha} + \Omega_{0}(\mathbf{\hat{n}}_{\alpha}\times\nabla p)\times\mathbf{\hat{n}}_{\alpha} +\sqrt{2D_{r}}\mathbf{\hat{n}}_{\alpha}\times\mathbf{f}_{\alpha}^{n}(t),$$

From coupled Langevin equations to a hierarchy of equations for the density field, the polarisation field, etc

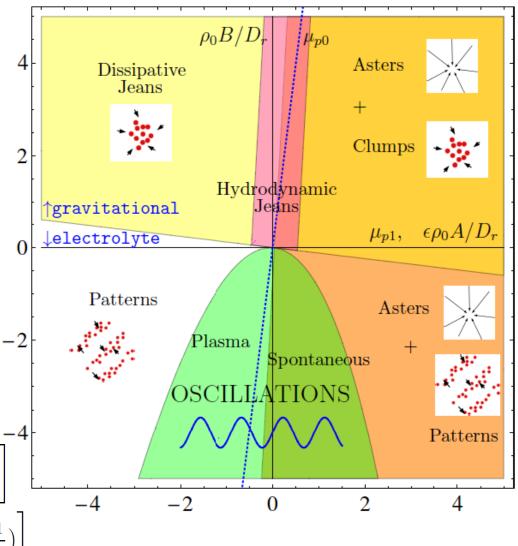
with all the coefficients known in terms of the microscopic parameters that can be tuned by changing the surface chemistry and geometry.

Collective Behaviour

Interplay between selfpropulsion, spin, drift, alignment, driven by and mediated via chemicals

- -> Variety of Behaviours:
 - Clustering
 - Aster condensation
 - Pattern formation
 - Plasma Oscillations
 - Ringing

$$A = N\kappa(s_0) \left[\frac{\Omega_0}{D_p} - \frac{\Phi_0}{D_s} + \frac{V_0(s_0)}{2D_s} \frac{d\ln\kappa}{ds} \Big|_{s_0} \right]$$
$$B = N\kappa(s_0) \left[\frac{1}{D_p} (\beta_0 + \frac{\beta_1}{3}) + \frac{1}{D_s} (\alpha_0 + \frac{\alpha_1}{3}) \right]$$



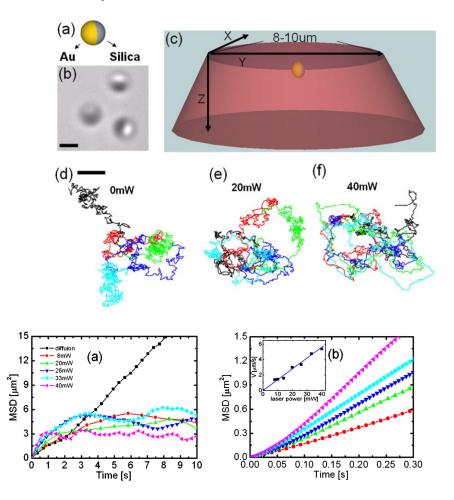
Part III

Thermal Activity

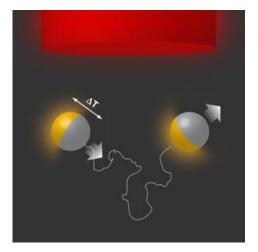
... self-organised phototactic swarms ...

Self-Thermophoresis

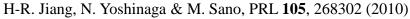
Experiment



Mechanism



R. Golestanian, T.B. Liverpool & A. Ajdari (2007)



Collective Thermotaxis

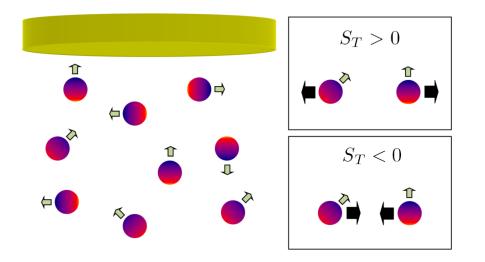
- Intra-colloid Activity
 - Mobile heat source
 - Self-propelled, if asymmetric

$$v_0 = \frac{\epsilon I D_T}{6\kappa}$$

 $\mathbf{v} = -D_T \nabla T$

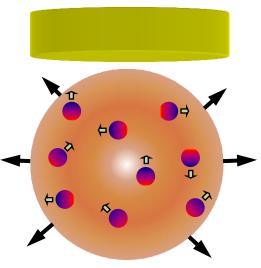
- Inter-colloid Interaction
 - Thermo-repulsive
 - Thermo-attractive

$$S_T = \frac{D_T}{D}$$



Thermo-Attractive ("Gravitational") Case





colloidal "star"

 Heat flux at the boundary cannot balance the heat generated inside
 -> uncontrolled build-up of heat

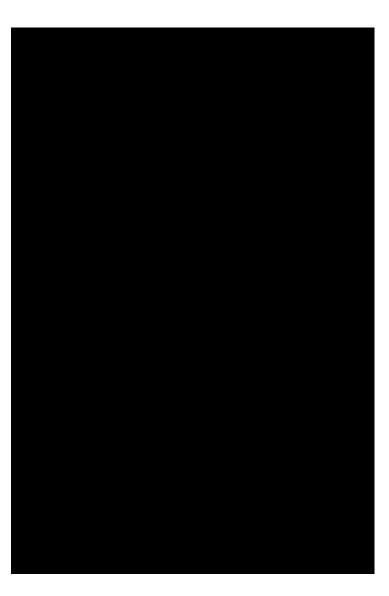


colloidal "supernova"

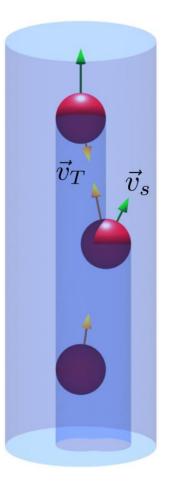
 In molecular systems: exothermic combustion reactions -> thermal explosion!

D.A. Frank-Kamenetskii, "Calculation of thermal explosion limits." Acta. Phys. Chin., URSS 10, 365-370 (1939)

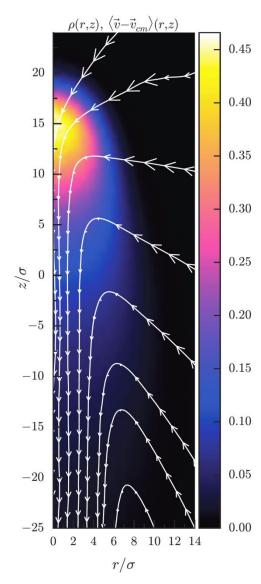
Collective Phototaxis



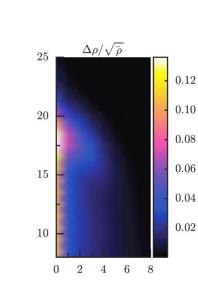
- Thermophoresis
- Attraction
- Excluded volume
- Shadowing
- uniform illumination
 from the top
- propulsion due to self-generated gradT
- mutual attraction due to gradT caused by mobile heat sources

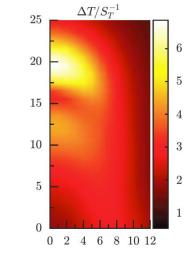


Comet-like Swarming



- uniform average motion
- hot-and-dense head
- circulation
- large density fluctuations & temperature fluctuations





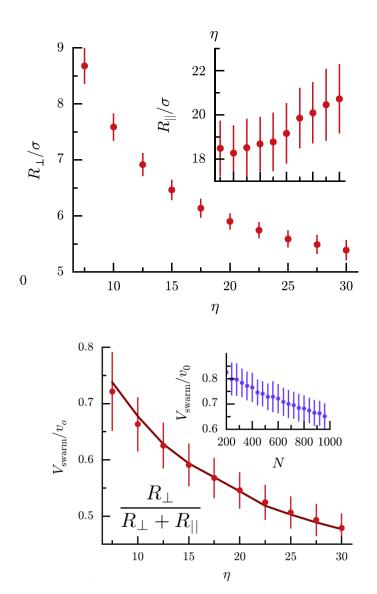
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Changing the Coupling Strength

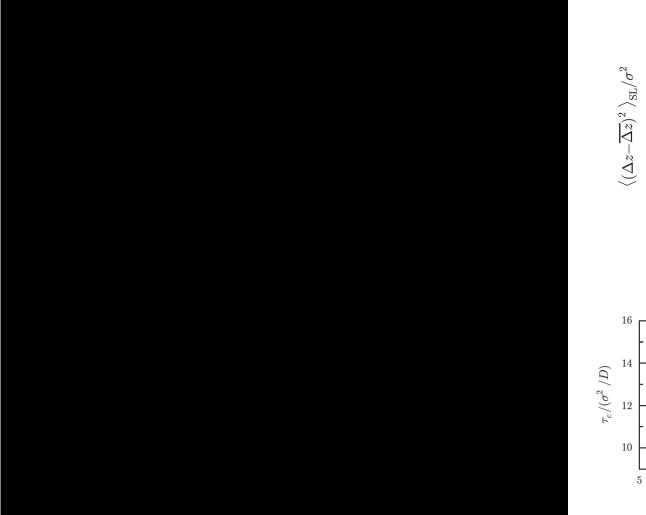
$$\eta = \sigma \epsilon I |S_T| / \kappa$$

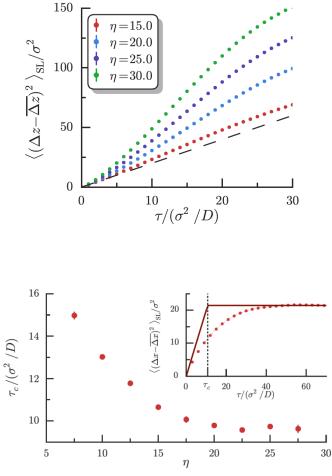
As the coupling strength increases, the Swarm:

- elongates
- narrows down
- slows down
- becomes more dynamic

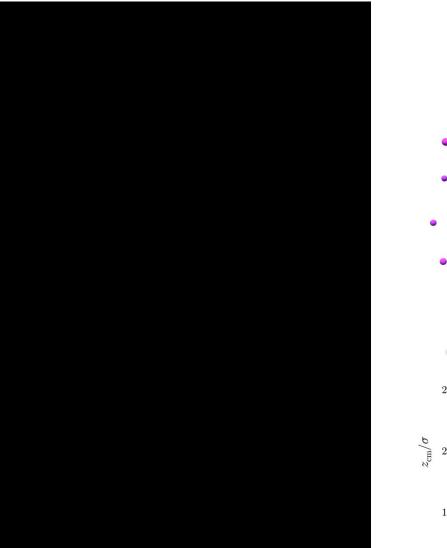


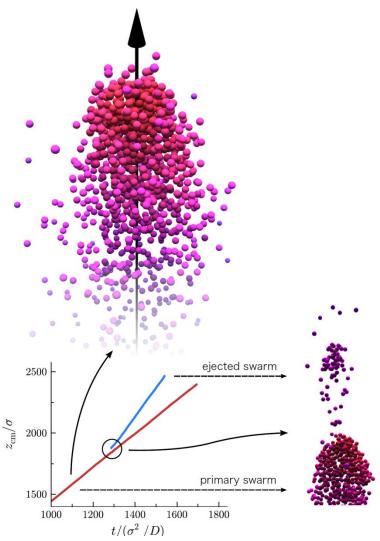
Particle Tracking: Super-diffusion & Confinement



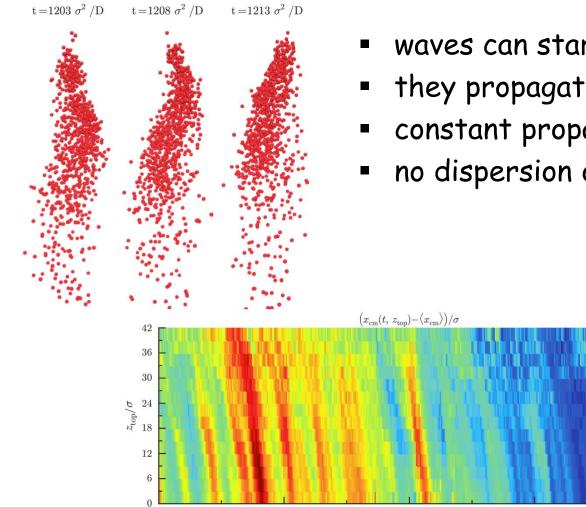


Increased Coupling Behaviour: Fission





Increased Coupling Behaviour: Waves



1100

1000

waves can start randomly

they propagate from tail to head

6

2

1400

- constant propagation velocity
- no dispersion or damping

1300



Conclusion

Collective behaviour of active colloids could be very rich due to the long-range nature of their **chemical** or **thermal** interactions.